

## Aromatic Hydroxylation: the Direct Oxidation of C–Pd into C–OPd

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The C–Pd bond in cyclopalladated azobenzenes [(1) and (3)] is converted into the C–OPd function [(2) and (5)] in good yield by stoichiometric reactions with *m*-chloroperbenzoic acid; in effect azobenzene is successively oxidised to 2-hydroxyazobenzene and 2,2'-dihydroxyazobenzene.

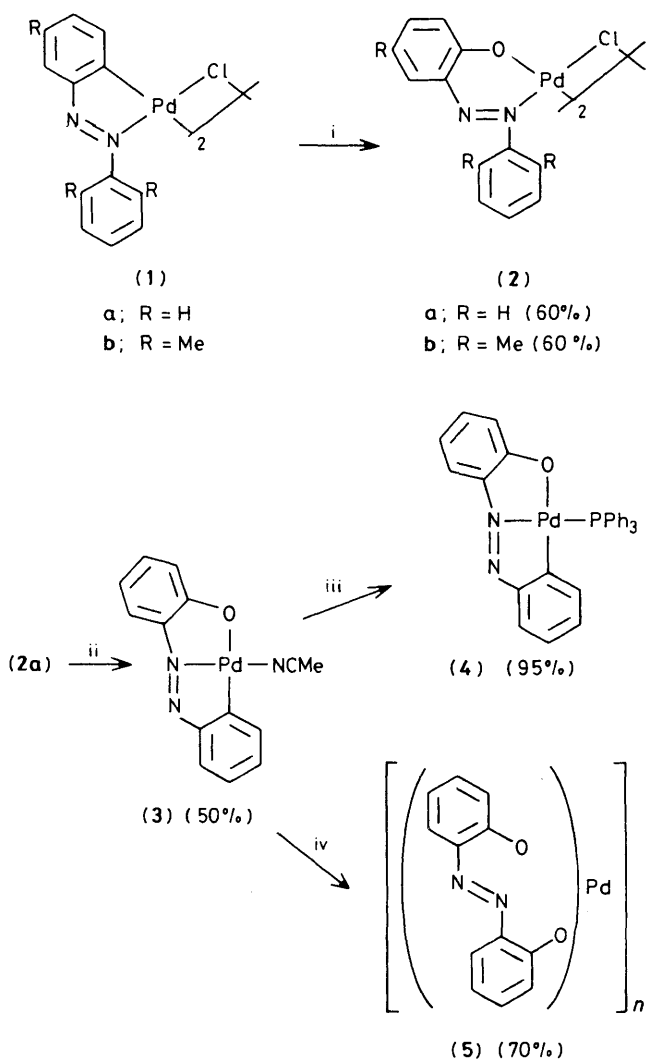
The hydroxylation of aromatic carbon atoms (C–H → C–OH) is a reaction of current interest.<sup>1</sup> We report here that the oxidation C–Pd → C–OPd can be directly achieved in palladated azobenzenes<sup>2</sup> (1) using *m*-chloroperbenzoic acid (*m*-CPBA) as the oxidant, Scheme 1. The details given for the conversion (1b) → (2b) are representative.

To an acetonitrile solution of (1b) (1 mmol) a solution of *m*-CPBA (3 mmol) in the same solvent was added dropwise at room temperature. After stirring for 3 h, the solvent was removed under reduced pressure and the residue was washed with aqueous ethanol and cold diethyl ether to remove excess of acid. Chromatography of a chloroform solution of the residue on a silica gel column with acetonitrile–benzene (5:95) as eluant gave complex (2b) which was isolated by evaporation under reduced pressure.

The various species shown in Scheme 1 are formed in good yields, and all except (3) have been isolated in pure solid form by simple crystallisation or by column chromatography on silica gel using benzene, acetonitrile, or a mixture of both as

eluants. These compounds have been characterized by elemental analysis, spectra, and, where possible, by unambiguous synthesis [(2b) and (5), see below] from preformed azophenols. Species (2) and (4) have  $\nu(\text{CO})$  ca. 1140 cm<sup>-1</sup>; (5) 1140 and 1120 cm<sup>-1</sup>;  $\lambda_{\text{max}}$  (CH<sub>2</sub>Cl<sub>2</sub>): (2a), 495 ( $\epsilon$  2 900 dm<sup>3</sup> mol<sup>-1</sup> cm<sup>-1</sup>); (2b), 495 (3 400); (4), 660 (6 000) and 625 (6 600); (5), 495 (9 800) and 450 (9 000) nm. The <sup>1</sup>H n.m.r. spectrum of (4) shows a doublet (1H, *J* 7.0 Hz) at  $\delta$  5.88; the remaining aromatic signals occur as a complex pattern in the range  $\delta$  6.3–7.7. As in the case<sup>3</sup> of the PPh<sub>3</sub> adduct of *ortho*-palladated azobenzenes, the unique high-field signal is assigned to the proton *ortho* to the metallated carbon in (4). This proton lies close to the phenyl groups of PPh<sub>3</sub> and the shift to high field arises owing to the shielding effect of these phenyl groups. In the pyridine analogue of (4) [synthesised as for (4), Scheme 1, but using pyridine in place of PPh<sub>3</sub>] all aromatic proton signals occur above  $\delta$  6.2.

Palladium(II) chloride reacts with 2',5,6'-trimethyl-2-hydroxyazobenzene in anhydrous ethanol to yield a reddish-



**Scheme 1.** Reagents and conditions: i, MeCN, *m*-CPBA; ii, adsorption on silica gel, 348 K, 2 h, MeCN; iii, CH<sub>2</sub>Cl<sub>2</sub>, PPh<sub>3</sub>; iv, MeCN, *m*-CPBA. Yields are based on the immediate precursors.

brown complex that is identical in all respects with (2b). This confirms that the C–OPd bond in (2b) arises directly from the C–Pd bond of (1b). Similarly, the reaction of 2,2'-dihydroxyazobenzene with Na<sub>2</sub>PdCl<sub>4</sub> in ethanol affords an orange-red complex that is identical to (5). Owing to its insolubility the molecular weight of (5) could not be determined; to date we have not been able to grow single crystals.

The *ortho*-metallation of (2a) (ii, Scheme 1) is achieved by adsorbing the complex on silica gel followed by mild heating of the dry material. On subsequent elution with MeCN a green solution of (3) is obtained from which MeCN can be displaced by donors such as PPh<sub>3</sub> [as in (4)], pyridine, etc. In this novel surface-induced *ortho*-metallation reaction the Si–OH groups<sup>4</sup> of the silica are thought to be active.

The mechanism and more general applicability of the C–Pd → C–OPd reaction are under study. The existing information<sup>5</sup> on oxidative cleavage of aliphatic M–C (M = Pd or Pt) species suggests that co-ordination of the oxidising agent is a possible initial step.

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